

# Raman signal enhancement of low-cost metal sheet SERS with gold decoration

Nampueng PANGPAIBOON<sup>1</sup>, Kawinphob PHETNAM<sup>1</sup>, Sukon KALASUNG<sup>2</sup>, Chanunthorn CHANANONAWATHORN<sup>2</sup>, Viyapol PATTHANASETTAKUL<sup>2</sup>, Mati HORPRATHUM<sup>2</sup>, Pitak EIAMCHAI<sup>2</sup>, Noppadon NUNTAWONG<sup>2</sup>, and Saksorn LIMWICHEAN<sup>2,\*</sup>

<sup>1</sup> Department of Industrial Physics and Medical Instrumentation, Faculty of Applied Science, King Mongkut's University of Technology North Bangkok, Bangkok, 10800, Thailand

<sup>2</sup> Opto-Electrochemical Sensing Research Team, Spectroscopic and Sensing Devices Research Group, National Electronics and Computer Technology Center, Pathum Thani, 12120, Thailand

Surface Enhance Raman Spectroscopy (SERS) is a sensitive surface creating plasmonic resonance

for the Raman scattering. SERS is considered as a powerful technique to enhance the Raman signal,

when quantity of sample is limited. Recently, our lab team has discovered a low-cost SERS template

fabrication, laser-engraving method, which is suitable with gold nanoparticles decoration. In this work,

two different materials, zinc, and copper, were used as metal sheet SERS substrate. Gold nanoparticles sputtered with 270 s of sputtering time were used as decorated noble metal. FE-SEM images illustrated the nano-in-microstructure of the engraved metal sheets. The decorated Au nanoparticles were uniform

and fully cover on the rough metal sheet templates. From contact angle measurement with DI water,

Zn/Au SERS provided the highest contact angle, which was  $142.10 \pm 0.51$  degree. SERS performance,

including Enhancement Factor (EF), Limit of Detection (LOD), and shelf-life were investigated. The EF

values of Zn/Au SERS and Cu/Au SERS were  $7.40 \times 10^9$  and  $1.70 \times 10^8$ , respectively. From LOD results,

Zn/Au SERS presented the great ability to enhance the Raman signal at 10<sup>-7</sup> molar concentration. However,

after 90 days of shelf-life test, both showed the capability to enhance the Raman signal.

\*Corresponding author e-mail: saksorn.limwichean@nectec.or.th

Abstract

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#### 1. Introduction

Surface Enhance Raman Spectroscopy (SERS) is one of the influential analytic innovations. Raman spectroscopy provides a characteristic fingerprint which can identify a molecule of substance. Nowadays, SERS is used as an ultra-sensitive surface for detecting chemical and biological species in many applications; for example, detecting herbicide in agriculture [1-3], and detecting addictive or explosive substance in forensic science [4-6]. The significant advantages of SERS, non-destructiveness, and fast sensing, make SERS become an interesting appliance for Raman spectroscopy. In principle, SERS can enhance the Raman signal by surface plasmon resonance (SPR). This phenomenon occurs at the surface of noble metal when freeelectrons cloud is vibrated by monochromatic light excitation. There are three types of SERS, including colloid base [7,8], thin-film base [9,10], and 3D-hybrid structure [3,11-15]. The conventional colloid base is not satisfied as a strong enhancement of Raman signal. This structure requires complex and high-cost process and produces non-repeatability device. For thin-film base structure, even it presents an ability to enhance the Raman signal strongly; however, thin film structure requires complex and high-cost process as same as the colloid base. Afterwards, 3Dhybrid structure is developed for reducing cost and simplify fabrication process. This structure is constructed with a nano-in-microstructure template decorated by the noble metal nanoparticles.

There are many methods to fabricate the 3D-hybrid structure SERS; for example, lithography [16-19], physical vapor deposition (PVD) [3,20-24], chemical vapor deposition (CVD) [25-27], hydrothermal [28-30] and laser marking [3,31-33]. From all these methods, laser marking or, laser engraving is an interesting one because of lowcost production, easy processing, and non-chemical waste technique. Recently, our research team has successfully fabricated SERS from an aluminum sheet by using laser engraving technique. We found that the laser-engraved aluminum sheet SERS decorated with Au nanoparticles presented high intensity and high stability Raman signal [3]. From the success, we are interested to discover a set of lowcost SERS template made by inexpensive material and present high efficiency as SERS device.

Considering the intended application and economic, zinc (Zn) has received much attention in industries due to its non-toxic and high electric performance. Additionally, zinc is the fourth most common metal use in production, just below of iron, aluminum, and copper [34]. For SERS application, zinc sheet has been reported to use as SERS substrate; for example, a novel and reproducible Zn/Au-Ag/Ag sandwich-structure SERS substrate discovered by Wang *et al.* 

in 2021 [35]. With the composite substrate, SERS demonstrated high sensitivity with a wide dynamic range because of a rough structure and its interfacial charge transfer. On the other hand, copper (Cu) substrate was proof that it is a good choice of SERS template because of highly efficient SERS platform with a low detection limit, good recyclability, and a long service time under ambient conditions [32]. In addition, copper has good biocompatibility, high conductivity, and chemical stability, which supports a great potential in ultra-trace detection of biomolecules.

Therefore, in this paper, two commercial low-cost metal sheets, zinc and copper, have been selected as a laser-engraved template for Au nanoparticles. The Raman signal enhancement of two lowcost metal sheet SERS were investigated. Roughness surfaces of these metal sheets were created by using a conventional laser-marking machine which is simple and inexpensive. Sputtering technique was used to decorate Au nanoparticles, SPR material, on the SERS templates. Topography of the SERS surface, topography of the decorating nanoparticles, hydrophobic property of the SERS and ability of the Raman signal enhancement were investigated. Moreover, efficiency of SERS was verified by limit of detection (LOD test), enhancement factor (EF), and shelf-life. A good performance SERS produced by low-cost metal sheets and inexpensive techniques was expected for this research.

## 2. Experimental

#### 2.1 Low-cost metal sheet SERS fabrication

Commercial zinc and copper sheets were selected to be an enhancing surface for the Raman signal. Two simple steps of metal sheet SERS fabrication are laser engraving, and noble metal decorating, as described by following. First, the clean zinc and copper sheet were engraved by a laser marking machine with wavelength 1064 nm (Smart mark, Photonics Science Co., Ltd). Laser power, fill spacing, and frequency used for laser engraving were 12 W, 0.02 mm, and 30 kHz, respectively. After that, the engraved zinc and copper sheets were coated with Au nanoparticles by magnetron sputtering system (lab-assembled, single-cathode PVD system). Base pressure for an ultra-high vacuum (UHV) sputtering system was  $5 \times 10^{-6}$  mbar. Au nano-particles deposition was performed at  $3.0 \times 10^{-3}$  mbar operating pressure for 270 s sputtering time. Cathode current and DC power for operating were 0.1 A and 90 W, respectively. Finally, all samples were cut into a square piece of  $5 \times 5 \text{ mm}^2$  dimension, attached on a glass slide, and sealed in metalized package. Then Au zinc sheet SERS (Zn/Au SERS) and Au copper sheet SERS (Cu/Au SERS)

were completely fabricated. The fabrication procedure of low-cost metal sheet SERS is shown in Figure 1.

#### 2.2 Characterization

Topography of bare zinc and copper sheet were examined through a field-emission scanning electron microscopy (FE-SEM). As same as metal sheet, FE-SEM was also used to characterize structure of Au nanoparticles on the metal sheets template. Diameter of the nanoparticles were measured by image J software. Hydrophobic property on the SERS substrate was analyzed by dropping deionized water (DI water), about 20 µL. on the center of SERS and measuring contact angles of DI water by contact angle machine. To verify an enhancement ability of Zn/Au SERS and Cu/Au SERS, Rhodamine 6G (R6G) with  $1 \times 10^{-5}$  molar concentration was used as a testing solution. Two milliliters of R6G were dropped on the SERS and left at room temperature for 30 min. After that, the Raman enhancements from Zn/Au SERS and Cu/Au SERS were examined by inVia Raman microscope from Renishaw. Laser wavelength, exposure time and magnification of Raman spectroscopy for the examination were 785 nm, 30 s and 5X, respectively. An ability to enhance the Raman signal at a very low concentration was also studied in this work by using limit of detection (LOD) test with R6G substance from  $1 \times 10^{-4}$  to  $1 \times 10^{-7}$  molar concentration. Enhancement factor (EF) for the Raman signal of Zn/Au SERS and Cu/Au SERS were calculated. Finally, shelf-life test was observed by keeping the SERS in an ambient environment for 90 days. The Raman signals were measured along the period.

# 3. Results and discussion

Zinc and copper sheet templates were completely fabricated by the laser marking method. Nano-in-microstructure of engraved zinc and copper sheets were investigated by FE-SEM, as shown in Figure 2(a)-(d). Figure 2(a) and (c) shows microstructure surface of bare zinc and copper sheets at 1 k magnification, and after engraving by laser marking, bare zinc sheet presents nano-in-microstructure at 100 k magnification, as shown in Figure 2(b). FE-SEM image illustrates a homogeneous rough surface and uniform pattern nano-in-microstructure of engraved zinc sheet. In the same way, after laser-engraved, the copper sheet surface is homogeneous. Through FE-SEM image exhibits non-uniform structure of engraved copper sheet, the nano-inmicrostructure is obviously presented in Figure 2(d). From the result, the engraved metal sheets, both Zn and Cu, are suitable for using as a supporting template for SERS.

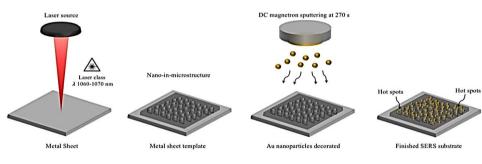


Figure 1. Schematic fabrication procedure for low-cost metal sheet substrate.

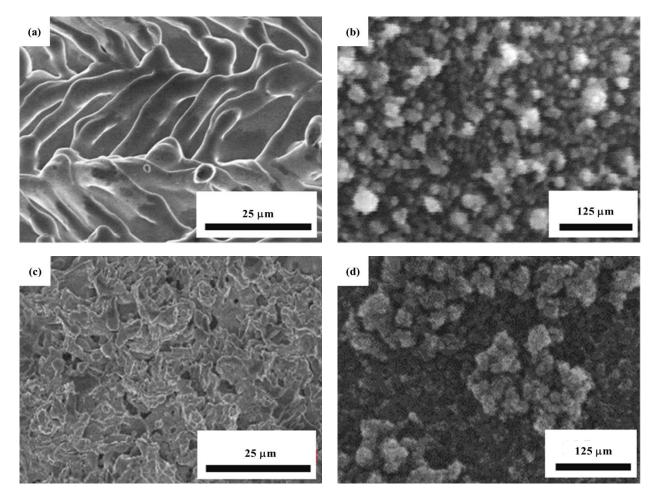


Figure 2. FE-SEM images illustrating by topography of bare zinc sheet and copper sheet (a) Bare zinc sheet after engraving at 1 k, (b) Bare zinc sheet after engraving at 100 k, (c) Bare copper sheet after engraving at 1 k, and (d) Bare copper sheet after engraving at 100 k.

After laser-engraved process, zinc and copper sheet templates were decorated with Au nanoparticle and then Zn/Au SERS and Cu/Au SERS were fabricated. Likewise, surface roughness of the engravedmetal sheets decorating with Au nanoparticles were investigated by FE-SEM. Topography images of Au nanoparticles on zinc and copper sheet templates are presented in Figure 3(a) and Figure 3(b), respectively. From FE-SEM, Au nanoparticles are spherical shape and formed as a cluster on the rough templates. The metal surfaces are fully covered by a uniform distribution of Au nanoparticles. Average diameters of Au nanoparticles on zinc and copper sheet template are  $44.87 \pm 4.64$  nm and  $55.28 \pm 6.03$  nm, respectively. Because of the nanoparticle's aggregation, as shown in FE-SEM images, hot-spot or inter gap between the nanoparticles which is commonly used to identify the performance of SERS cannot be measured. Therefore, to predict enhancing ability of Zn/Au SERS and Cu/Au SERS, some characterization techniques were demonstrated in this part.

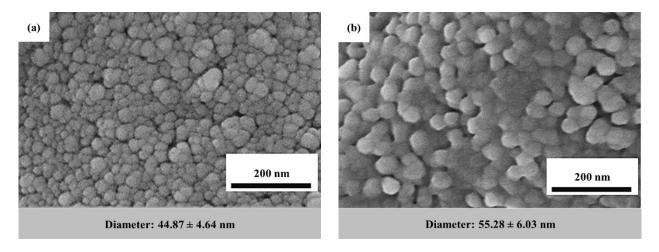


Figure 3. FE-SEM images illustrating by topography of Au nanoparticles decorated on (a) zinc sheet template, and b) copper sheet template.

From literature reviews, hydrophobic property of SERS substrate can estimate an efficiency to enhance the Raman signal of SERS [12]. We found that contact angles of Zn/Au SERS and Cu/Au SERS are 142.1  $\pm$  0.51 degree and 131.3  $\pm$  0.41 degree, respectively, as shown in Figure 4(a)-(b). The higher contact angle of Zn/Au SERS, comparing with Cu/Au SERS, refers to probability of a strong Raman signal enhancement. Since small contact area between solution droplet and SERS substrate presents high concentration of a sample solution. After solution dry, molecule of the solution will settle at the bottom, in the small contact area. Therefore, this area will fully condense with molecule of solution, which increases an intensity of the Raman signal.

The Raman spectrums of R6G testing solution on Zn/Au SERS and Cu/Au SERS are shown in Figure 5. Both of Zn/Au SERS and Cu/Au SERS can identify the characteristic Raman signal fingerprint of R6G molecule with the Raman shift at 610 cm<sup>-1</sup>, 1367 cm<sup>-1</sup>, and 1512 cm<sup>-1</sup>. From Raman spectroscopy analysis, the Zn/Au SERS reveals higher intensity of the Raman signal than the Cu/Au SERS according to the results of hydrophobic test from the previous section. Comparing the Raman signal enhancement of our SERS with the commercial SERS chip (Onspec Nectec LITE SERS chip, Al/Au SERS), we found that the Zn/Au SERS shows the highest Raman intensity, following by the Al/Au Onspec SER chip and the Cu/Au SERS, respectively. For application, one of the important properties of SERS is an ability to enhance the Raman signal at a very low concentration solution. Therefore, the lowest concentration of testing solution that SERS can enhance the Raman signal were studied, which is called limit of detection (LOD) testing. LOD of Zn/Au SERS and Cu/Au SERS are tested by using  $1 \times 10^{-4}$  molar to  $1 \times 10^{-8}$  molar concentration of R6G solution. Figure 6(a) and (b) present LOD results of R6G Raman spectrum from Zn/Au SERS and Cu/Au SERS. The Raman shift of R6G Raman spectrum can be observed by the Zn/Au SERS until  $1 \times 10^{-7}$  molar concentration. After that, at  $1 \times 10^{-8}$  molar concentration, the Zn/Au SERS cannot provide the R6G Raman shift. On the other hand, the Cu/Au SERS can provide the R6G Raman shift until  $1 \times 10^{-6}$  molar concentration.

To investigate an enhancing efficiency of our SERS, enhancement factor (EF value) was calculated. EF value is widely used to report SERS performance [36]. This value represents an ability of SERS to enhance the Raman signal at any concentration of solution. EF value can be calculated by Equation (1).

$$EF = \left(\frac{I \, surf}{I \, bulk}\right) \times \left(\frac{N \, bulk}{N \, surf}\right) \tag{1}$$

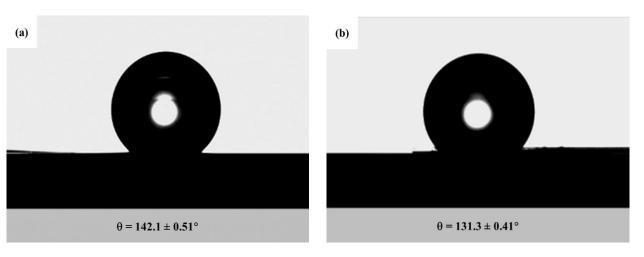


Figure 4. Contact angles of DI water droplet on (a) Zn/Au SERS, and (b) Cu/Au SERS.

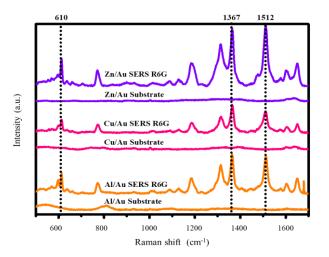


Figure 5. Raman spectra of R6G solution enhanced by Al/Au SERS, Cu/Au SERS and Zn/Au SERS.

As in the equation, Isurf is an intensity of the vibration modes on the SERS. Ibulk is the intensity of vibration mode from the Raman spectrum. N<sub>bulk</sub> is the number of a molecules on the Raman spectrum. Nsurf is the number of a probed molecules that upon the SERS surface. For this research, EF values of Zn/Au SERS, Cu/Au SERS, and the commercial Al/Au SERS (Onspec SERS chip) were calculated by using the Raman shift of R6G Raman spectrum at 1367 cm<sup>-1</sup> as a reference peak. The EF values of Zn/Au and Cu/Au SERS are  $7.40 \times 10^9$ and  $1.70 \times 10^8$ , respectively. Zn/Au SERS shows higher EF value than Cu/Au SERS as expected. Moreover, comparing with the commercial Al/Au SERS, the EF value of our low-cost Zn/Au SERS is noticeably greater than of the commercial SERS, which is  $4.13 \times 10^9$ . Besides, from literatures, EF values have been reported in the range of  $5.49 \times 10^6$ to  $4.20 \times 10^7$  [36-39]. Therefore, the low-cost metal sheet substrate, Zn/Au SERS and Cu/Au SERS present the impressive ability to enhance the Raman signal.

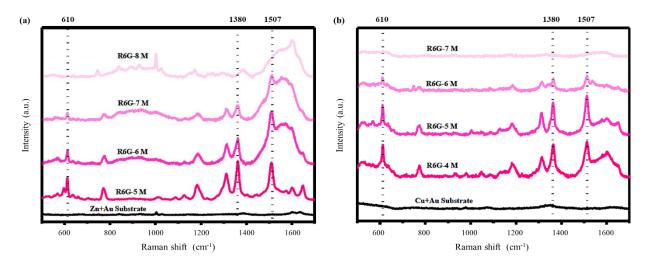


Figure 6. Limit of detection (LOD) of R6G solution from a) Zn/Au SERS and b) Cu/Au SERS.

The last examination, shelf-life test reveals an ability to enhance the Raman signal of SERS while keeping in an ambient environment for a long time. The intensities of the Raman signal were measured frequently along that period. SERS capability can be de-graded by three major factors: humidity, temperature, and pressure. In this experiment, shelf-life test of Zn/Au SERS and Cu/Au SERS were observed and compared with the commercial Al/Au Onspec SERS chip. R6G solution at  $1 \times 10^{-5}$  molar concentration was used as a testing solution. Data of the enhanced Raman signal intensities were collected for 90 days, and the results are presented in Figure 7. The intensities of R6G Raman spectrum relating with shelf-life (days) were calculated by using the average intensities of the reference peaks at 610 cm<sup>-1</sup>, 1367 cm<sup>-1</sup>, and 1512 cm<sup>-1</sup> of R6G Raman spectrum. Before shelflife testing, the Zn/Au SERS presents the highest enhanced intensity of the Raman spectrum, following by the commercial Al/Au SERS and the Cu/Au SERS, and this trend remains for the first 7 days. During 14 days to 30 days, the Raman intensity from the Zn/Au SERS obviously decreases while the Raman intensity of the Cu/Au SERS turns to be the highest order. The Cu/Au SERS still presents higher Raman signal intensity than the Zn/Au SERS and the commercial Al/Au SERS for 90 days of our test. From the results, we found that both Zn/Au SERS and Cu/Au SERS have considerably efficiency to enhance the Raman signal. The Zn/Au SERS has a significant ability to enhance the Raman signal while the Cu/Au SERS presents a valuable long shelf-life advantage.

In practical using, SERS may be destroyed by many factors making SERS degradation, in this experiment, we try to observe the effects in the same way as normal use. Therefore, shelf-life properties were tested in an ambient environment. However, for suggestion, to increasing service life of SERS, the SERS should be stored in nitrogen-filled metalized bags as well as be controlled humidity and temperature during utilizing process.

From this research, we successfully fabricated low-cost metal substrate SERS, zinc and copper sheet, by using the simple and inexpensive laser engraving technique, which can reduce cost of production and improve the performance of SERS. The engraved zinc and copper sheets were sputtered with Au nanoparticles creating the surface plasmon resonance. The electron clouds from Au located on the low-cost metal sheet perfectly amplify the Raman signal. We found that Zn/Au SERS and Cu/Au SERS reveal a good ability to enhance the Raman signal comparing with the literatures and the commercial one. From LOD testing and EF value, the Zn/Au SERS presents the highest efficiency as expected. It can enhance the Raman signal even when the concentration less to  $1 \times 10^{-7}$  molar concentration for R6G. The great performance of the Zn/Au SERS is a result of the homogeneous and uniform roughness surface. Moreover, the decorating Au nanoparticles are well-constructed with the engraved zinc nano-in-microstructure. This result harmonizes with the hydrophobic properties of the Zn/Au SERS which presented the high contact angle of DI water.

On the other hand, the Cu/Au SERS exhibits the remarkable result. Capability to enhance the Raman signal of the Cu/Au SERS remained constant until 90 days, which suitable for in general application. Since copper is a noble metal as same as Au, the electron clouds on the copper surface can cooperate with the electron clouds of Au nanoparticles. Then two materials mutually amplify the SPR mechanism which causes the performance of the Raman enhancement.

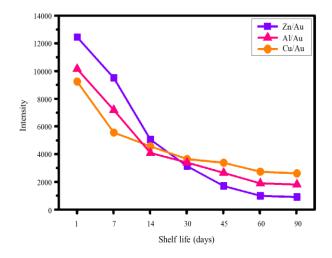


Figure 7. Relationship between shelf-life (days) and intensity of R6G Raman spectrum of Zn/Au SERS, Al/Au SERS, and a commercial SERS chip (Onspec Nectec LITE SERS chips, Al/Au SERS).

### 4. Conclusions

Low-cost metal SERS chips, Zn/Au SERS and Cu/Au SERS were successfully fabricated by laser-engraved technique. Topography images studied by FE-SEM reveal nano-in-microstructure of the engraved templates and the fully covered Au nanoparticles on these metal sheet surfaces. Hydrophobic property of the SERS surface can predict an ability to enhance the Raman signal. Performance of Zn/Au SERS and Cu/Au SERS were studied by measuring enhancement Raman intensity, calculating EF values, testing LOD and observing shelf-life. The Zn/Au SERS presents the high value of EF and better LOD results. Although the Cu/Au SERS shows longer period usability, both SERS are suit for general application and their performances are greater than the commercial SERS chip.

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